

The Atomic Theory as Applied To Gases, with Some Experiments on the Viscosity of Air

by

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B.S. Mechanical Engineering, UCLA, 2018

M.S. Stellar Interiors, Vulcan Science Academy, 2020

MBA, Ferengi School of Management, 2022

Submitted to the Department of Physics and the Department of Mechanical Engineering in partial fulfillment of the requirements for the degrees of

BACHELOR OF SCIENCE IN PHYSICS

and

BACHELOR OF SCIENCE IN MECHANICAL ENGINEERING

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ABSTRACT

The developments of the “kinetic theory” of gases made within the last ten years have enabled it to account satisfactorily for many of the laws of gases. The mathematical deductions of Clausius, Maxwell and others, based upon the hypothesis of a gas composed of molecules acting upon each other at impact like perfectly elastic spheres, have furnished expressions for the laws of its elasticity, viscosity, conductivity for heat, diffusive power and other properties. For some of these laws we have experimental data of value in testing the validity of these deductions and assumptions. Next to the elasticity, perhaps the phenomena of the viscosity of gases are best adapted to investigation¹.

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¹Text from Holman (1876): doi:[10.2307/25138434](https://doi.org/10.2307/25138434).